

# **Quiz 4 Metal Reactivity**

### Quiz set by Ms. T. wint for Year 10/10Y/Sc3

- [] salt and hydrogen
- [] alkali and oxygen

### 2) The test for a particular gas is a squeaky pop. This gas is...

- [] hydrogen
- [] helium
- [ ] nitrogen
- []oxygen

# 3) Which of the following metals will not burn in oxygen to form a metal oxide?

- []gold
- [] calcium
- [ ] zinc
- [ ] iron

#### 4) Which metal will react vigorously with dilute acid?

- [] magnesium
- [ ] zinc
- [ ] iron
- []copper

### 5) What is the state symbol for aqueous solutions?

• [](aqu)

<ul><li>[](a)</li><li>[](sol)</li><li>[](aq)</li></ul>
6) When sodium reacts with hydrochloric acid, which gas is produced?
<ul><li>[] hydrogen</li><li>[] carbon dioxide</li><li>[] oxygen</li></ul>
7) What are the products when calcium reacts with nitric acid?
<ul> <li>[ ] Calcium oxide and carbon dioxide</li> <li>[ ] Calcium chloride and hydrogen</li> <li>[ ] copper nitrate and hydrogen</li> <li>[ ] Calcium nitrate and hydrogen</li> </ul>
8) A metal that can be used for making water pipes is
<ul><li>[] lithium</li><li>[] copper</li><li>[] iron</li></ul>
9) Group 1 metals are very reactive with water
• [] True • [] False
10) When potassium reacts with water, Hydrogen is produced and
<ul> <li>[ ] Copper Sulfate</li> <li>[ ] Potassium Oxide</li> <li>[ ] Potassium Sulfate</li> <li>[ ] Potassium Hydroxide</li> </ul>



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